NATIVE ELEMENTS IN METEORITES AND CONTINENTAL LITHOSPHERE

Abstracts of All-Union Conference "Native elements formation in the endogenic processes"
Yakutsk, June 25-28, 1985

art IV

METAL PORMATION AND FRACTIONATION OF SIDEROPHILE ELEMENTS IN THE SOLAR NEBULA

Gero Kurat (x), Herbert Palme (xx), Ernst Pernicke (xxx), Franz Brendstätter)(x), Bernard Spettal (xx), Ingrid Herrwerth (xxx), and V.P. Perelygin (xxxx).

(x) Naturhistorisches Museum, A-1014 Vienna, Austria.
(xx) Max-Flanck-Institut für Chemie, D-65 Mainz, FRG.
(xxx) Max-Flanck-Institut für Kernphysik, D-69 Heidelberg, FRG.
(xxxx) Joint Institute for Nuclear Research, Dubna, USSR.

chondritic rock) appears to be unfractionated. This situation, chondrites reveals severe fractionations of siderophile elein (or sometimes outside) the solar system. Metal in different ferent conditions of formation and thus different places withmental fractionations and isotopic inhomogeneities observed. closed systems. On the other hand, the sometimes extreme elenaturally, indicates an evolution of chondrites within fairly ments within each component and yet, the total mixture (= the reality, this situation places close constraints on the condinon-volatile lithophile elements) appear to be miraculous. In dy generally contain abundant metal and have non-volatile sideamong some chondritic constituents strongly indicate very dif-In addition, close inspection of the different constituents of tions of formation of these rocks early in the solar nebula. tionated abundances of the siderophile elements (and also the chaotic, microbreccia-like rocks (e.g. 3,4), the almost unfrac-(1,2). Considering the fact that chondrites are very complex, rophile element relative abundances similar to solar matter Chondrites, the oldest and most primitive rocks we can stu-

constituents of chondrites as well as total siderophile element contents record a variety of formation conditions and processes.

Metal occurs in chondrites (mostly together with sulfides)
(a) within chondrules and lithic fragments either as droplets or as "dusty" inclusions in silicates,

- (b) as rims around chondrules and fragments.
- (c) as metal-sulfide chondrules,
- (d) as aggregates within the chondrite matrix, and
- (e) in small emounts in highly fractionated "inclusions" (mainly Ca-rich, Al-rich) (see 5).

metal under reducing conditions. te). On the other hand, siderophile elements (Ni,Co) originally oxidized (formation of phosphates and oxides) or sulfurized nisms exist : (a) Sampling, (b) Separation of silicate melts dissolved in silicates under oxidizing conditions exsolved as ons, metal formed originally under reducing conditions becomes Commonly, metal in chondrules records changing redox conditihighly variable Ni/Co ratios indicate oxidizing conditions. tion processes. Variable contents of lithophile elements in meelements relative to each other. Solid/liquid-vapor fractionadocumented by a variety of fractionations of the siderophile nal evaporation. Chondrule bulk data show that most of these from metal melts, (c) Fractional condensation, and (d) Fractioseparating siderophile elements from lithophiles was very effeccan be as low as 0.005 (8-11). Thus, a fractionation mechanism Depletion factors between 0.1 and 0.3 are very common but they commonly strongly depleted in siderophile elements (e.g. 6,7). (formation of troilite or pyrrhotite, sometimes even pentlandital (Si, Or, P) record reducing conditions. High Ni metals and processes were operative during chondrule formation. This is tively operating in the early solar nebula. Four possible mechations are common and testify for both evaporation and condensa-Chondrules, now mostly crystalline once molten droplets, are

Metal and siderophile element contents of chondrite components thus testify for widely differing redox conditions prevailing in the early solar nebula. They further record complex processing of individual chondrite components and exposure to

very different and changing f02, f52 and T conditions. No systematic sequences for changing conditions can be established now, except for a late stage oxidizing event which appears to be very common and which probably established the final oxidation grade of the ordinary chondrites (H, L and LL groups).

LENCES

Russell, H.N. (1939) Astrophys. J.70, 11-82.

Suess, H.E. (1949) Zeitschr. Elektrochemie 53, 237-241.

Kurst G. (1984) Proc 27th Internat. Geol. Congr., vol. 11, 155-197.

Kurat G. (1975) Tschermaks Min. Petr. Mitt. 22, 38-78.

El Goresy, A. (1985), this volume.

Howard, E. (1802) Phil. Trans. Royal Soc. London, I, 168-212. Dodd, R.T. (1978) Barth Planet. Sci. Letts. 29, 52-66.

Grossman, J.N. and Wasson, J.T. (1982) Geochim. Cosmochim. Acta $\underline{46}$, 1081-1099.

Pernicka, E., Kurat, G., Brandstätter, G., and Herrwerth, I. (1985) in preparation.

Grossman, J.N. and Wasson, J.T. (1985) Geochim. Cosmochim.Acta, in press.

Kurat, G., Palme, H., Brandstätter, F., Spettel, B., and Perelygin, F.P. (1985), in preparation.

MINERAL INCLUSIONS IN DIAMONDS FROM THE SLOAN KIMBERLITES, COLORADO

Henry O.A. Meyer and Malcolm E. Mc Callum

Dept. of E Geosciences, Purdue University, West Lafayette, In 47907, Dept of Earth Resources, Colorado State University, Cort Collins, CO 80521

Mineral inclusions have been recovered from about 40 small (1.5 mm) diamonds from the Sloan 1 and 2 kimberlite pipes in Northern Colorado. Protogenetic and/or syngenetic minerals are olivine, clinopyroxene, sanidine, rutile and possibly native olivine, chinopyroxene, sanidine, rutile and possibly native Fe and some phlogopite. Phases interpreted as epigenetic are acmite, richterite, most phlogopite, perovskite, Mn-ilmenite, Cr-spinel, magnetite, clacite, serpentine and possibly been reported in diamond. Epigenetic minerals commonly occur as in-